

# Formation process of Al<sub>2</sub>O<sub>3</sub> thin film by reactive sputtering

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Reactive sputtering is one of the widely used techniques to prepare compound thin films. In this study, Al and Al<sub>2</sub>O<sub>3</sub> films were deposited by RF magnetron sputtering using mixed gas of Ar and O<sub>2</sub>. The formation process of the Al<sub>2</sub>O<sub>3</sub> films was studied by plasma emission spectroscopy and target voltage measurements. Al<sub>2</sub>O<sub>3</sub> films with transmittance above 95% and refractive index of 1.50-1.66 were found to be formed at O<sub>2</sub> flow ratios above 8%. Plasma emission peaks of O atoms (777nm) were clearly seen at oxygen flow ratios above 8%, and target voltage decreased drastically at O<sub>2</sub> flow ratios above 8%. It is thought that the surface of Al target was changed from metallic mode into oxide mode at the O<sub>2</sub> flow ratios of 8%. The relationship between the amount of sputtered Al atoms and the amount of supplied O<sub>2</sub> molecules was studied, and the atomic ratio of sputtered Al atoms to supplied oxygen atoms was found to be approximately 2:3 at the critical O<sub>2</sub> flow ratio of 8%.

## 1. Introduction

Al<sub>2</sub>O<sub>3</sub> films have attracted much attention for optical, and electrical applications due to their high dielectric constant and high mechanical, chemical and thermal stabilities.

Reactive sputtering is one of the most common techniques used for obtaining compound thin films. Al<sub>2</sub>O<sub>3</sub> films have also been prepared by reactive sputtering and their formation process has been studied [1, 2]. However, quantitative discussions on the formation process of Al<sub>2</sub>O<sub>3</sub> films and the target surface oxidation have not been reported in detail. The purpose of this study is to clarify the influence of O<sub>2</sub> flow ratio, which is necessary to form Al<sub>2</sub>O<sub>3</sub> thin films, and the influence of O<sub>2</sub> flow ratio on the oxidation of an Al target. Optical and electrical properties of deposited films, plasma and target surface states were studied as a function of O<sub>2</sub> flow ratio.

## 2. Experimental

Al and Al<sub>2</sub>O<sub>3</sub> films were deposited by RF (13.56 MHz) magnetron sputtering, using an Al (99.99% purity) target with 2-in diameter. Corning #7059 glass and Si were used as substrates. Mixed gas of Ar and O<sub>2</sub> was used for sputtering and the O<sub>2</sub> flow ratio was varied from 0% to 100%. Total gas pressure, total gas flow rate, RF input power, and substrate temperature were fixed at 5mTorr, 5cc/min, 50W, and room temperature, respectively. The thickness of the deposited films was about 100nm. The flow rates of Ar and O<sub>2</sub> gases were controlled by mass flow controllers.

The crystal structure of the films was evaluated by X-ray diffraction (XRD) with Cu K $\alpha$  radiation. Electrical resistivity was measured by a four-point probe method. Transmittance was examined by a multi-channel charge-coupled device (CCD) detector. Refractive index of the films was examined by ellipsometry with a He-Ne laser (633nm). Plasma condition during sputtering was evaluated by plasma

emission spectroscopy and target surface state was checked by target self-bias voltage. The amount of Al atoms sputtered from the Al target was estimated from the mass of deposited films.

### 3. Results and discussion

First, effects of O<sub>2</sub> flow ratio on the formation of Al<sub>2</sub>O<sub>3</sub> films were studied. Transmittance of the films at a wavelength λ of 600nm is plotted as a function of O<sub>2</sub> flow ratio in Figure 1. The transmittance is very low for the films deposited at O<sub>2</sub> flow ratios of 0-6%, and it increases to 95-100% at O<sub>2</sub> flow ratios above 8%. The refractive index of the transparent films at a wavelength of 633nm was 1.50-1.66, which agrees well with the value of Al<sub>2</sub>O<sub>3</sub> (1.54-1.78) [3, 4]. The resistivity of the film was about 4μ cm at an O<sub>2</sub> flow ratio of 0% and it increased to 18μ cm at that of 6%. The resistivity increased too high to measure by the four point probe method at oxygen flow ratios above 8%. XRD peaks due to Al(111) was observed for the films deposited at O<sub>2</sub> flow ratios of 0-6%, however, amorphous films were formed at 8-100%. From these results it is thought that Al films were formed at O<sub>2</sub> flow ratios below 8%, and Al<sub>2</sub>O<sub>3</sub> films were formed at O<sub>2</sub> flow ratios above 8%.

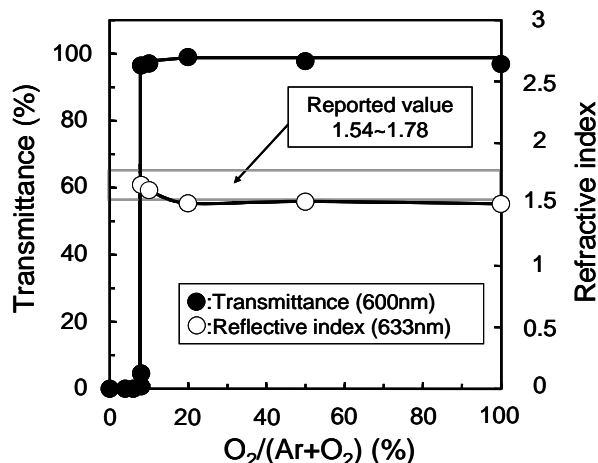


Figure 1. Transmittance (λ=600nm) and refractive index (λ=633nm) of the films as a function of O<sub>2</sub> flow ratio.

Plasma emission intensity of oxygen atoms [5] is shown as a function of O<sub>2</sub> flow ratio in Figure 2. From this figure, it is seen that the emission intensity of oxygen atoms is very small at O<sub>2</sub> flow ratios below 8%, and it begins to increase at 10%. It is thought that oxygen molecules are gettered by Al atoms deposited on the sputtering chamber wall and the substrate, and the oxygen density in the plasma are very low at O<sub>2</sub> flow ratios of 0-8%. Due to the very low density of oxygen in the plasma, Al films are formed. Above a critical O<sub>2</sub> flow ratio of 8% the amount of supplied O<sub>2</sub> molecules is supposed to exceed the gettering effect and oxygen density in the plasma begins to increase. As a result Al<sub>2</sub>O<sub>3</sub> films are formed. At the critical O<sub>2</sub> flow ratio of 8%, the amount of supplied O<sub>2</sub> molecules was calculated to be  $1.8 \times 10^{-5}$  mol/min. And the amount of sputtered Al atoms was estimated to be  $2.3 \times 10^{-5}$  mol/min from the mass of the deposited Al film. This means that the ratio of the number of sputtered Al atoms to that of supplied oxygen atoms becomes the stoichiometric ratio of 2:3, at the critical O<sub>2</sub> flow ratio.

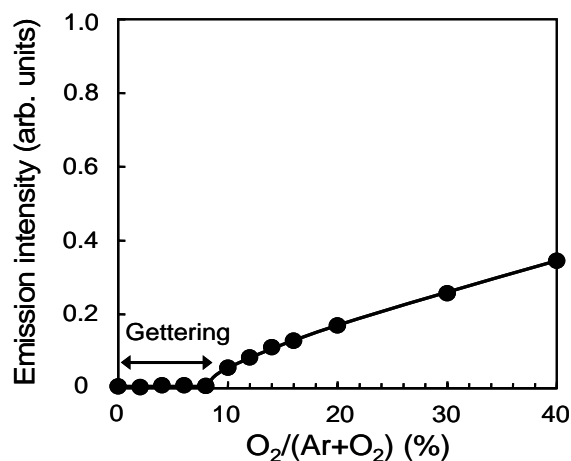


Figure 2. Plasma emission intensity of oxygen atoms (λ=777nm) as a function of O<sub>2</sub> flow ratio.

Figure 3 shows the change of the deposition rate as a function of O<sub>2</sub> flow ratio. It is clearly seen that the deposition rate decreases abruptly above the critical O<sub>2</sub> flow ratio of 8%. Figure 4 shows the change of

the target self-bias voltage as a function of O<sub>2</sub> flow ratio. The target voltage also decreases abruptly above the critical O<sub>2</sub> flow ratio of 8%. These phenomena are attributed to a change of the target mode as shown in Figure 5. The formation process of Al and Al<sub>2</sub>O<sub>3</sub> films is classified into the following two regions (a) metal target mode region at oxygen flow ratios below 8%, where Al films were formed at a high deposition rate of about 14nm/min, (b) oxide target mode region at oxygen flow ratios above 8%, where Al<sub>2</sub>O<sub>3</sub> films were formed at a low deposition rate of about 1nm/min.

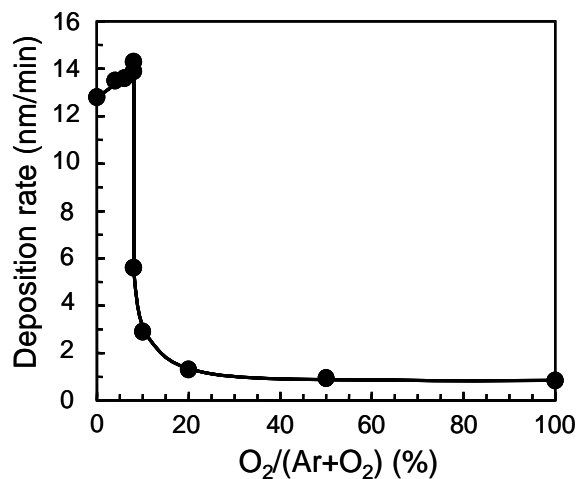


Figure 3. Deposition rate of the films as a function of O<sub>2</sub> flow ratio.

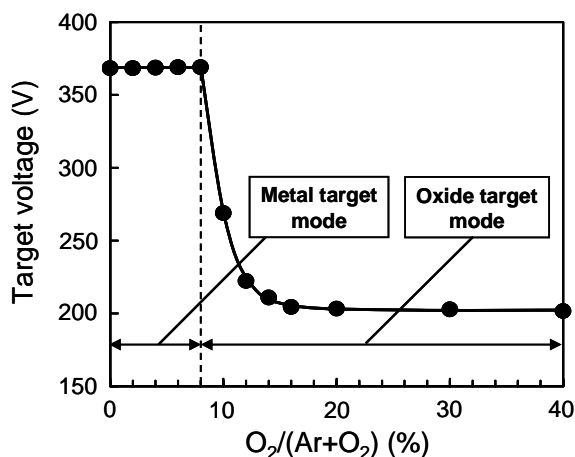


Figure 4. Target voltage as a function of O<sub>2</sub> flow ratio.

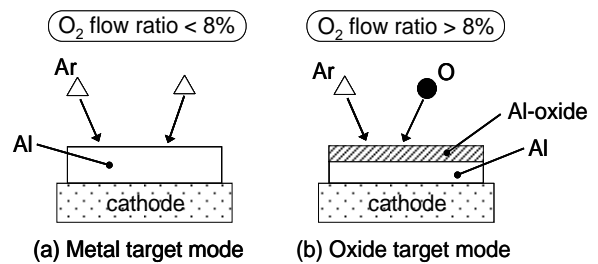


Figure 5. A model of target mode change.

#### 4. Conclusion

The formation process of Al and Al<sub>2</sub>O<sub>3</sub> films was classified into two regions; (a) metal target mode region, and (b) oxide target mode region. At a critical O<sub>2</sub> flow ratio of 8%, where the target mode changed from metal mode to oxide mode, it was found that the ratio of the amount of sputtered Al atoms to that of supplied oxygen atoms became the stoichiometric ratio of 2:3. Al films were formed in the metal target mode region, because oxygen molecules supplied into the sputtering chamber were gettered and oxygen density in the plasma was very low. Above the critical O<sub>2</sub> flow ratio, the amount of supplied O<sub>2</sub> molecules exceeded the gettering effect and oxygen density in the plasma began to increase. As a result Al<sub>2</sub>O<sub>3</sub> films were formed.

#### REFERENCES

- [1] S. Maniv and W. D. Westwood : J. Appl. Phys. 51 (1980) 718.
- [2] D. Depla, J. Haemers, G. Buyle and R. De Gryse : J. Vac. Sci. Technol. A24 (2006) 934.
- [3] H. K. Pulker : Coatings on Glass (Elsevier, Amsterdam, 1999) 2nd ed., p.409.
- [4] E. D. Palik ed. : Handbook of Optical Constants of Solids (Academic Press. San Diego, 1998) p.676.
- [5] D. R. Lide ed.: Handbook of Chemistry and Physics (CRC Press, Boca Raton, 2003) 84th ed., p.10-53.