

Durability of Lightweight Concrete and Mortar Exposed under Some Environment

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(Received April 30, 1991)

Abstract

In this experiment, identically composed specimens of lightweight concrete and of a mortar material were exposed to a cold (coastal)* environment (Monbetsu), warm sea environment (Izu) and used in the roof of a building in a warm-climate city (Yokohama) for 3years. The physical and chemical properties of the lightweight concrete specimens and those of the mortar specimens were examined. The results are as follows:

- (1) The lightweight concrete specimen contained more salt than the same mortar specimen.
- (2) The free calcium content on the surface of lightweight concrete specimens tended to decrease more than that of the mortar specimen.
- (3) Ettringite by sulfuric ion and Friedel's salt by chlorine ion were not observed in either the lightweight concrete specimens or the mortar specimens.

1. INTRODUCTION

Experiments were carried out in such a way that specimens of lightweight concrete were exposed in cold coastal environment (Monbetsu), warm sea environment (Izu) and warm town area's roof of building (Yokohama) for 3years. The physical and chemical properties of the specimens of lightweight concrete and the same water cement ratio and materials were inspected.

2. EXPERIMENTAL PROCEDURES

Mixing condition and material are shown in Table 1 and Table 2. Exposed conditions are shown in Table 3. The method of measurement are shown in Table 4.

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Table 1 Mix condition and properties of concrete

Maximum size of coarse aggregate (mm)		W/C (%)		S/a (%)	
15		45		45	
Mixing condition (Kg/m ³)					
Water	Cement	Fine aggregate	Coarse aggregate	Chemical admixtures 1	Chemical admixtures 2
156	347	802	585	0.868	2.086
Properties					
Slump (cm)			Air content (%)	Temperature of fresh concrete (°C)	
Base concrete	After adding HP-20				
11.6	22.8		6.35	20.3	

Table 2 Material

Kinds of cement	Fine aggregate	Coarse aggregate	Chemical admixtures 1	Chemical admixtures 2
Blast-furnace slag cement B(S.G.: 3.06)	S.G.: 2.62 A.R.: 1.50%	S.G.: 1.60 (Light-weight)	Air entrained	Super-plasticizer

S.G.: Specific gravity A.R.: Absorption rate (%)

Table 3 Exposing condition

Exposing Spot	Distance from sea water	Temperature (°C)	Relative humidity	Cycles of freeze and thaw
Monbatsu	32m	Average minimum of Feb.: -11.3 Average maximum of Aug.: 22.3 Average of year: 5.9	Av. of Feb.: 76 Av. of Aug.: 85 Av. of year: 76	80
Izu	1m	Average minimum of Feb.: 1.8 Average maximum of Aug.: 30.5 Average of year: 16.0	Av. of Feb.: 58 Av. of Aug.: 78 Av. of year: 69	20
Yokohama	10km	Average minimum of Feb.: 1.2 Average maximum of Aug.: 30.7 Average of year: 15.1	Av. of Feb.: 60 Av. of Aug.: 80 Av. of year: 71	20
Standard curing		20	—	0

*1: Referring to Tokyo Astronomical Observatory: "Science chronological table"

Table 4 Items of test and dimension of specimens

Items of test	Dimension of specimen (cm)	Test method
Weight	Concrete $\square 10 \times 10 \times 40$	By balance (sensitivity : 1g) measured immediately and measured after soaking for 4 hours
	Concrete $\phi 10 \times 20$	
	Mortar $\square 10 \times 10 \times 20$	
Resonance frequency	Concrete $\square 10 \times 10 \times 40$	The same as the above condition in accordance with JIS A 1127
	Concrete $\phi 10 \times 20$	
Ultra sonic velocity	Concrete $\square 10 \times 10 \times 40$	The same as the above condition by using ultrasonic pulse apparatus
	Mortar $\square 10 \times 10 \times 20$	
Compressive strength	Concrete $\phi 10 \times 20$	The same as the above in accordance with JIS A 1107

3. CONCLUSION

(1) Physical properties

Relation between period of exposing and change in weight are shown in Fig. 1. Relation between period of exposing and change in resonance frequency are shown in Fig. 2. Relation between period of exposing and change in ultrasonic velocity are shown in Fig. 3. Relation between period of exposing and change in compressive strength are shown in Fig. 4.

- Example
- : Monbetsu, measured immediately
 - : Mombetsu, measured after soaking for 4 hours
 - : Izu, measured immediately
 - : Izu, measure after soakig for 4 hours
 - ▲ : Yokohama, measured after soaking for 4 hours
 - △ : Yokohama, measured after soaking for 4 hours
 - ◇ : Standard curing

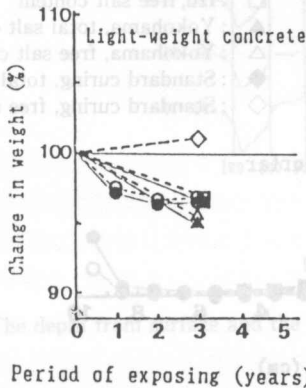


Fig. 1 Relation between period of exposing and change in weight

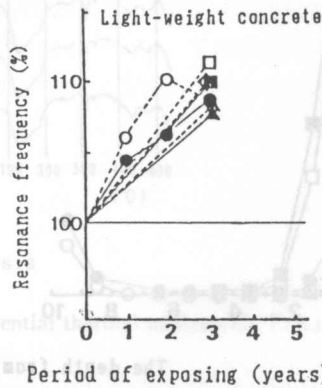


Fig. 2 Relation between period of exposing and change in resonance frequency

4.

Accordingly physical properties of the specimens after exposing in some environment were nearly the same as that of the specimens after standard curing and there was hardly any deterioration observed in specimens.

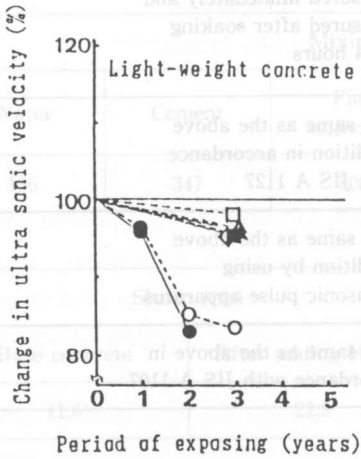


Fig. 3 Relation between period of exposing and change in ultra sonic velocity

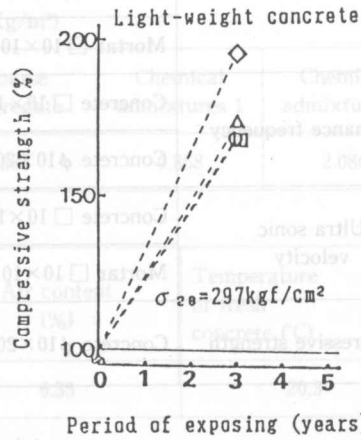


Fig. 4 Relation between period of exposing and change in compressive strength

(2) Chemical properties

Relation between the depth from the surface and salt content are shown in Fig. 5. The lightweight concrete specimen contained more salt than the same mortar.

Relation between the depth from the surface and free calcium content are shown in Fig. 6.

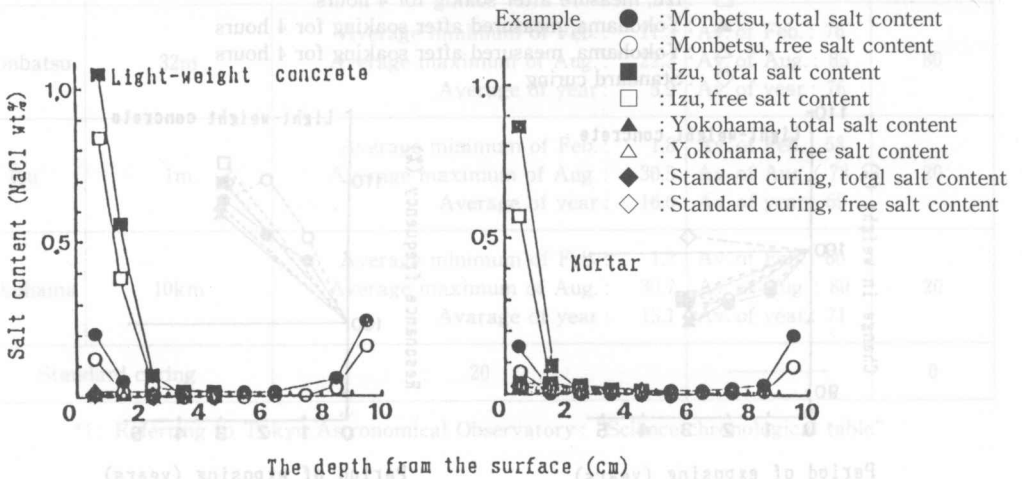


Fig. 5 Relation between the depth from the surface and salt content exposed in Monbetsu.

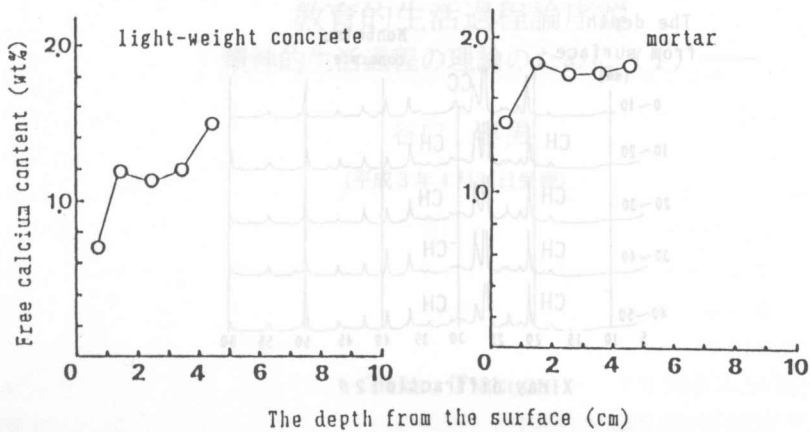


Fig. 6 Relation between the depth from the surface and free calcium content exposed in Monbetsu.

The depth from surface and the result of differential thermal analysis (D. T. A.) are shown in Fig. 7.

The depth from surface and the result of X-ray diffraction (X. R. D.) test are shown in Fig. 8.

Ettringite by sulfuric ion and Friedel's salt by chloride ion were not observed in both lightweight concrete and the mortar.

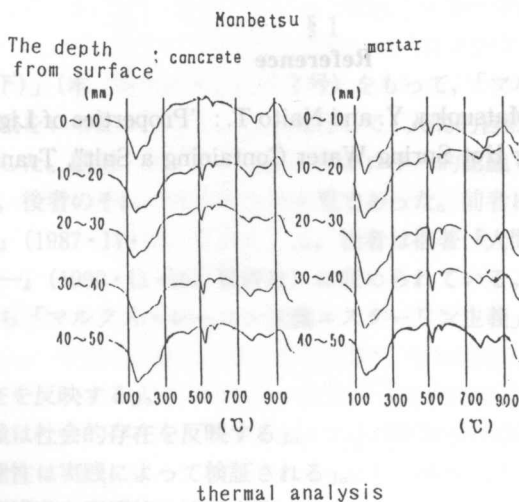


Fig. 7 The depth from surface and the result of differential thermal analysis (D. T. A.)

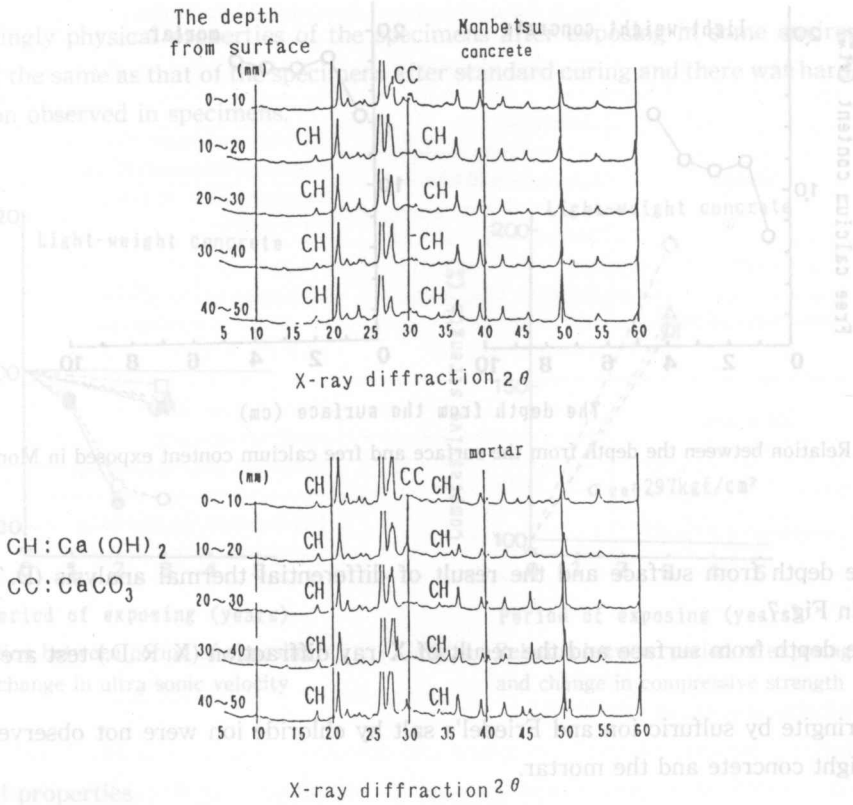


Fig. 8 The depth from surface and the result of X-ray diffraction (X. R. D.) test

Reference

1) Sakurai, H., Kaneko, S., Matsuoka, Y. and Naito T. : "Properties of Lightweight Concrete Soaked in Sewage Including Hot Spring Water Containing a Salt", Transaction of JCI vol. 7, 1985