

Degradation mechanism of hydrodynamic cavitation reactor

Shengqiang Li¹, Jian Wang², Ye Liu³, Longfeng Ji⁴, Chang Yang⁵

¹Wuhan University of Science and Technology
(No.2,Huangjiahu Road,Hongshan District, Wuhan 430065, Hubei, P.R. China)
Telephone: +86 18671842741
Email: 1176700543@qq.com

²Wuhan University of Science and Technology
(No.2,Huangjiahu Road,Hongshan District, Wuhan 430065, Hubei, P.R. China)
Telephone: +86 18171422510
E-mail: okwj@wust.edu.cn

³Wuhan University of Science and Technology
(No.2,Huangjiahu Road,Hongshan District, Wuhan 430065, Hubei, P.R. China)
Telephone: +86 15623189989
Email: 853983246@qq.com

⁴Wuhan University of Science and Technology
(No.2,Huangjiahu Road,Hongshan District, Wuhan 430065, Hubei, P.R. China)
Telephone: +86 15826567387
E-mail: 2453288826@qq.com

⁵Wuhan University of Science and Technology
(No.2,Huangjiahu Road,Hongshan District, Wuhan 430065, Hubei, P.R. China)
Telephone: +86 15272839007
E-mail: 1369751399@qq.com

Hydrodynamic cavitation is a promising technology in the degradation of the residual antibiotics which can exist in the environment for long periods as they are chemically stable. This study reviews the application of hydrodynamic cavitation to the residual antibiotics treatment. Varied methods for degrading antibiotics, bubble dynamics models coupled with chemical reactions, experiment setup, and cavitation generators are evaluated. A novel self-excited cavitation reactor which can produce pulse and cavitation using the fluid characteristic is proposed. Compared with traditional hydrodynamic cavitation generator, the self-excited pulsed cavitation jet generator has a longer hydraulic detention time, faster bubble collapse, more hydroxyl radical production, better treatment effect, and wider application prospects. Simple design, high energy efficiency, and no secondary pollution contribute to the great potential of self-excited cavitation reactor in wastewater treatment.

Key Words: hydrodynamic cavitation, hydroxyl radicals, pulsed cavitation generator, degradation rate, antibiotics

1. INTRODUCTION

The antibiotics, which has been utilized for more than 70 years, can be used not only against human and animal bacterial infections but also as additives for animal growth¹. While the residual antibiotics can exist in the environment for long periods as they are chemically stable and not easy

for biodegradation. They may result in the generation of antibiotics resistance genes (ARGs) as well as antibiotics resistance bacteria (ARB), even causing microbial resistance by horizontal gene transfer among pathogen organisms which may infect humans².

The traditional treatment strategies in the world include Physical treatment³⁻⁵ (coagulation-sedimentation method, adsorption method, membrane separation method, etc.), Chemical

treatment⁶⁻⁹(ozone oxidation method,Fenton oxidation method, photocatalytic oxidation method, electric-chemical oxidation method, etc.),and biological treatment^{10,11}(activated sludge method, biological contact oxidation method,anaerobic sludge bed method, etc.).Adsorption is a mature method as physical treatment but the regeneration of adsorbent is difficult¹².Chemical treatment require high investment and advanced equipment, and tend to bring about secondary pollution for the addition of chemical agent.Biological treatment is not a useful method for antibiotics for its inherent antibacterial property.Reverse osmosis and nanofiltration can remove the antibiotics from wastewater effectively,while the high running cost and membrane pollution restrict their widely application.Electrocatalytic oxidation technology with high efficiency and no secondary pollution property,consumes a lot of energy^{13,14}.Hence, a kind of simple, energy saving, high efficiency, environmentally friendly technology for treatment of antibiotics in wastewater is urgent and attractive.

Hydrodynamic cavitation(HC),as an advanced oxidation process with advantages of simple reactor design and no secondary pollution,becomes a hotspot for scholars from all over the world,but the low efficiency and high energy consumption hinder the industrial application of HC.

In this review,the hydrodynamic cavitation to the residual antibiotics treatment is studied.A novel hydrodynamic cavitation device called self-excited cavitation reactor is proposed.Different aspects self-excited cavitation reactor for wastewater treatment are evaluated.The emphasis is focused on both the related theoretical development and the experiment design and optimization.Some important contributions are incorporated,and their advantages and disadvantages are also compared.Some useful recommendations for both antibiotics degradation and self-excited application are proposed to benefit the industrial application.

2. HYDRODYNAMIC CAVITATION DEVICE

2.1 Traditional hydrodynamic generators

Hydrodynamic cavitation is a special advanced oxidation technology that has been used in wastewater treatment for about 20 years.The collapse of the bubble can provide a very extreme environment with high temperature(1,000~10,000K),and high-pressure(100-5,000bar) as well as high-speed micro-jet,the water molecules trapped in the bubble

will be decomposed into high reactive free radicals such as HO •, H •, HOO •, HO₂ •, H₂O₂ etc.¹⁵.Among the free radicals, HO • has extremely powerful oxidizability which can degrade the majority of organic pollutants with chain reaction,and decompose the organic pollutants into CO₂, H₂O,and minerals non-selectively without secondary pollution.

According to the academic papers published, the research hotspots could be summarized into three aspects.

(1) Operational parameters optimization

Operational parameters play important roles on the effect of hydrodynamic cavitation.Generally,these parameters include the inlet pressure,the temperature,the pH,the initial pollutant concentration and the initial radius of nuclei.

Under 3.0 bar inlet pressure for 10 mg/L initial concentration of RhB solution at 40 °C operating temperature in the presence of Fe³⁺-doped TiO₂ with 0.05:1.00 M ratio of Fe and Ti,the best HC degradation ratio can be obtained (91.11%)¹⁶.Optimum conditions for the maximum cell disruption using HC were cavitation device orifice over venturi, time (180 min),pressure (5 bar),and solid load (0.45% w/v)¹⁷.The maximum bacterial disinfection and COD reduction were obtained using CN5 reactor operated at 5 bar pump discharge pressure and the extent of reduction was found to be 59.17% and 52% respectively¹⁸.Although all these researches use the venturi cavitation device,different geometries and configurations require various optimum conditions due to diverse removal mechanisms¹⁹.

Meanwhile,even for the same pollutant,different cavitation devices require different operational parameters.Based on venturi HC device,the optimal operation parameters in HC was obtained: initial concentration of 30 μmol·L⁻¹, inlet pressure of 0.4 MPa,temperature of 25 °C²⁰.The hybrid scheme treatment based on HC by using Ecowirl reactor (20 °C, 2.0 bar, pH 2.0 or 4.0) and NaOCl (minimal dose 0.5 mg·L⁻¹) was found to be the most energy efficient and environmental friendly method to treat wastewater containing RhB²¹.For orifice plates, the highest cavitation yield can be obtained at pH 3 and initial dye concentration of 10 mg·L⁻¹.Also,an increase in the inlet pressure would lead to an increase in the extent of decolorization (ED)²².

As a result,we need to adjust the operational parameters according to specific cavitation device and specific pollutants.

(2) Coupled with other advanced oxidation processes

In situ generation of Fenton reagents can induce an advanced Fenton process (AFP) which can enhance the effect of the used catalyst in the HC process. Using a pilot scale hydrodynamic cavitation reactor, with iron metal blades as the heterogeneous catalyst, for catalytic decontamination of unsymmetrical demethylhydrazine (UDMH) wastewater, the highest cavitation yield can be obtained at pH 3 and initial UDMH concentration of 10 mg/l. In addition, the optimum value of 3 bar was determined for the downstream pressure that resulted to 98.6% degradation of UDMH after 120 min of processing time²³. In the work of removing p-nitrophenol, two different cavitating devices viz. orifice plate and venturi have been used. Effect of different operating parameters such as initial concentration (5 g/l and 10 g/l), inlet pressure (over a range 5.7-42.6 psi) and pH (over a range 2-8) on the extent of removal has been investigated. In conventional Fenton process two loadings of FeSO₄, 0.5 g/l and 1 g/l were investigated and three ratios of FeSO₄:H₂O₂ viz. 1:5, 1:7.5 and 1:10 were used. Removal observed with venturi was higher than with orifice plate in combination with Fenton chemistry. For 5 g/l initial concentration of p-nitrophenol, maximum removal of 63.2% was observed whereas for 10 g/l solution it was 56.2%²⁴.

In general, coupled with AOP technologies can bring higher efficiency and wider degradation extent. While complicated operating parameters and secondary pollutants bring about new problems.

(3) Exploration of novel cavitation generator

Generally, the cavitation generators can be classified into two categories, the linear cavitation and the rotating cavitation. Linear cavitation is conventional includes venturi tube generator, orifice plate generator, liquid whistle reactor et al.

Petkovšek et al. proposed a novel rotating cavitation generator shown in Fig.1 which uses the opposite movements of two shear layers to generate shear cavitation²⁵. This kind of cavitation generator has larger cavitation volume, rapid pressure recovery and lower pressure loss compared to the linear cavitation device. Ranade et al. disclosed a novel vortex-based cavitation device for effluent treatment which offered cavitation yield that was 3 to 8 times better (milligrams of pollutant degraded per Joule) compared to that from the orifice plate²⁶.

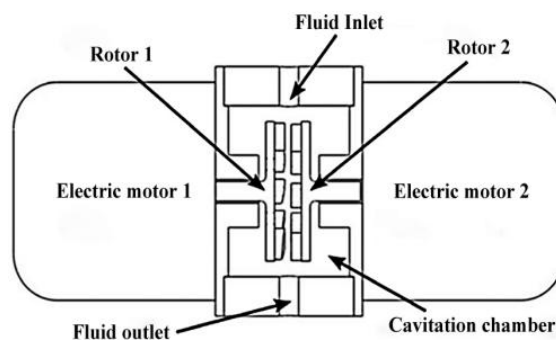


Figure.1 Rotating cavitation generator

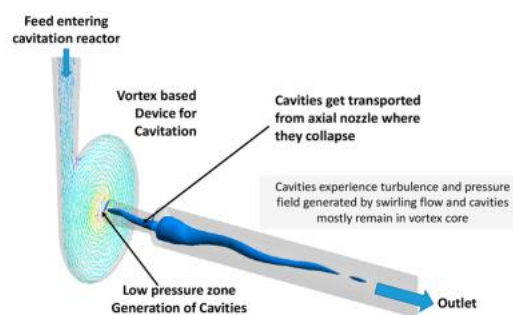


Figure.2 Vortex diode cavitation generator

2.2 Self-excited cavitation reactor

Self-excited cavitation reactor is a novel cavitation generator. Using the unique chamber structure and related operating parameters, this device can change the continuous energy into pulsed one and improve the hitting force significantly²⁷. As a result, it offers more violent collapse to the bubbles and generate more hydroxyl radicals and improve the degradation rate of the antibiotics or any other refractory organic pollutants.

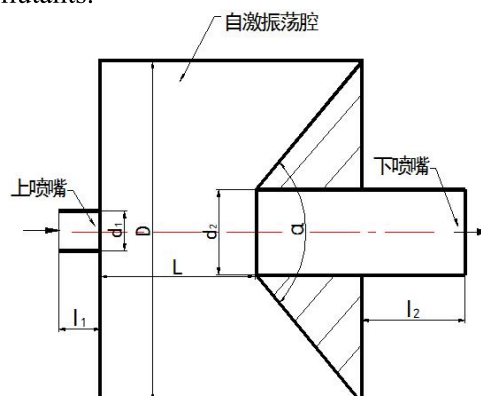


Figure.3 The structure of self-excited cavitation reactor

The working process of the self-excited cavitation reactor can be divided into two stages: energy storage and energy release as is shown in Fig.4. The simulated formation of the cavity is shown in Fig.5.

The self-excited cavitation generating technology is promising for its high efficiency, simple design and no by-products. While the modelling of the inception, growth, movement and collapse of the bubble is a big challenge.

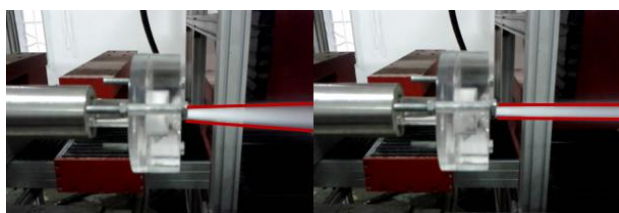


Figure.4 Working process of self-excited cavitation reactor

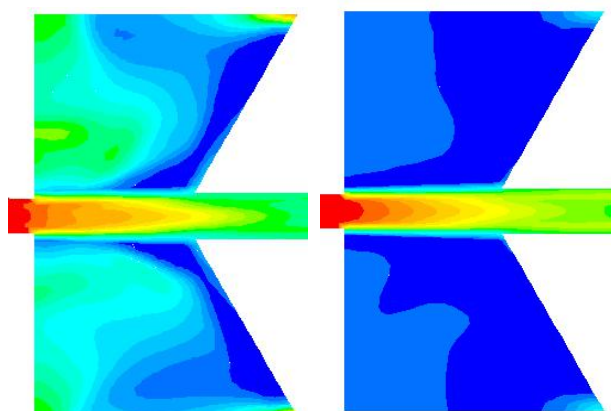


Figure.5 The formation of cavity in the chamber

Table1 the advantage of self-excited cavitation reactor

	Energy consumption	hydraulic detention time	bubble collapse speed	hydroxyl radical production
self-excited reactor	less	long	fast	more
traditional hydrodynamic generators	more	short	slow	less

3. CONCLUSION

Hydrodynamic cavitation is a promising technology for refractory organic wastewater treatment as it can provide a very special environment with high temperature and high pressure and generate hydroxyl radicals with strong oxidation. In the present proposal, we review the conventional cavitation generator and rotating cavitation generator, compare the pros and cons of every kind of the device. Thereafter, we propose a novel pulsed cavitation generator, illustrate the working principle, list the advantages and point out the challenge of modelling.

ACKNOWLEDGMENTS

This work was financially supported by “Science and Technology Research Project of Hubei Provincial Department of Education(Q20171107)”.

REFERENCES

- 1) ZHANG Q-Q, YING G-G, PAN C-G, et al. Comprehensive Evaluation of Antibiotics Emission and Fate in the River Basins of China: Source Analysis, Multimedia Modeling, and Linkage to Bacterial Resistance[J]. *Environmental Science & Technology*, 2015, 49(11): 6772-6782.
- 2) WANG J, ZHUAN R, CHU L. The occurrence, distribution and degradation of antibiotics by ionizing radiation: An overview[J]. *Science of the Total Environment*, 2019, 646: 1385-1397.
- 3) SAITOH T, SHIBATA K, FUJIMORI K, et al. Rapid removal of tetracycline antibiotics from water by coagulation-flotation of sodium dodecyl sulfate and poly(allylamine hydrochloride) in the presence of Al(III) ions[J]. *Separation and Purification Technology*, 2017, 187: 76-83.
- 4) ZHANG X, GUO W, NGO H H, et al. Performance evaluation of powdered activated carbon for removing 28 types of antibiotics from water[J]. *Journal of Environmental Management*, 2016, 172: 193-200.
- 5) DOLAR D, GROS M, RODRIGUEZ-MOZAS S, et al. Removal of emerging contaminants from municipal wastewater with an integrated membrane system, MBR-RO[J]. *Journal of Hazardous Materials*, 2012, 239: 64-69.
- 6) LIU Y, FAN Q, WANG J. Zn-Fe-CNTs catalytic in situ generation of H₂O₂ for Fenton-like degradation

- of sulfamethoxazole[J]. *Journal of Hazardous Materials*, 2018, 342: 166-176.
- 7) LIU H, PU C, YU X, et al. Removal of tetracyclines, sulfonamides, and quinolones by industrial-scale composting and anaerobic digestion processes[J]. *Environmental Science and Pollution Research*, 2018, 25(36): 35835-35844.
 - 8) TANG J, WANG J. Metal Organic Framework with Coordinatively Unsaturated Sites as Efficient Fenton-like Catalyst for Enhanced Degradation of Sulfamethazine[J]. *Environmental Science & Technology*, 2018, 52(9): 5367-5377.
 - 9) BELTRAN F J, AGUINACO A, GARCIA-ARAYA J F, et al. Ozone and photocatalytic processes to remove the antibiotic sulfamethoxazole from water[J]. *Water Research*, 2008, 42(14): 3799-3808.
 - 10) GUO R, CHEN J. Application of alga-activated sludge combined system (AASCS) as a novel treatment to remove cephalosporins[J]. *Chemical Engineering Journal*, 2015, 260: 550-556.
 - 11) FENG L, CASAS M E, OTTOSEN L D M, et al. Removal of antibiotics during the anaerobic digestion of pig manure[J]. *Science of the Total Environment*, 2017, 603: 219-225.
 - 12) KIM H, HWANG Y S, SHARMA V K. Adsorption of antibiotics and iopromide onto single-walled and multi-walled carbon nanotubes[J]. *Chemical Engineering Journal*, 2014, 255: 23-27.
 - 13) GIRALDO A L, ERAZO-ERAZO E D, FLOREZ-ACOSTA O A, et al. Degradation of the antibiotic oxacillin in water by anodic oxidation with Ti/IrO₂ anodes: Evaluation of degradation routes, organic by-products and effects of water matrix components[J]. *Chemical Engineering Journal*, 2015, 279: 103-114.
 - 14) OTURAN N, WU J, ZHANG H, et al. Electrocatalytic destruction of the antibiotic tetracycline in aqueous medium by electrochemical advanced oxidation processes: Effect of electrode materials[J]. *Applied Catalysis B-Environmental*, 2013, 140: 92-97.
 - 15) GOGATE P R. Hydrodynamic Cavitation for Food and Water Processing[J]. *Food and Bioprocess Technology*, 2011, 4(6): 996-1011.
 - 16) LI G, YI L, WANG J, et al. Hydrodynamic cavitation degradation of Rhodamine B assisted by Fe³⁺-doped TiO₂: Mechanisms, geometric and operation parameters[J]. *Ultrasonics sonochemistry*, 2019, 60: 104806-104806.
 - 17) WAGHMARE A, NAGULA K, PANDIT A, et al. Hydrodynamic cavitation for energy efficient and scalable process of microalgae cell disruption[J]. *Algal Research-Biomass Biofuels and Bioproducts*, 2019, 40.
 - 18) DOLTADE S B, DASTANE G G, JADHAV N L, et al. Hydrodynamic cavitation as an imperative technology for the treatment of petroleum refinery effluent[J]. *Journal of Water Process Engineering*, 2019, 29.
 - 19) TAO Y, CAI J, HUAI X, et al. Application of Hydrodynamic Cavitation to Wastewater Treatment[J]. *Chemical Engineering & Technology*, 2016, 39(8): 1363-1376.
 - 20) YI C, LU Q, WANG Y, et al. Degradation of organic wastewater by hydrodynamic cavitation combined with acoustic cavitation[J]. *Ultrasonics Sonochemistry*, 2018, 43: 156-165.
 - 21) MANCUSO G, LANGONE M, LAZZA M, et al. Decolourization of Rhodamine B: A swirling jet-induced cavitation combined with NaOCl[J]. *Ultrasonics Sonochemistry*, 2016, 32: 18-30.
 - 22) PARSA J B, ZONOZIAN S A E. Optimization of a heterogeneous catalytic hydrodynamic cavitation reactor performance in decolorization of Rhodamine B: Application of scrap iron sheets[J]. *Ultrasonics Sonochemistry*, 2013, 20(6): 1442-1449.
 - 23) ANGAJI M T, GHIAEE R. Decontamination of unsymmetrical dimethylhydrazine waste water by hydrodynamic cavitation-induced advanced Fenton process[J]. *Ultrasonics Sonochemistry*, 2015, 23: 257-265.
 - 24) PRADHAN A A, GOGATE P R. Removal of p-nitrophenol using hydrodynamic cavitation and Fenton chemistry at pilot scale operation[J]. *Chemical Engineering Journal*, 2010, 156(1): 77-82.
 - 25) PETKOVSEK M, ZUPANC M, DULAR M, et al. Rotation generator of hydrodynamic cavitation for water treatment[J]. *Separation and Purification Technology*, 2013, 118: 415-423.
 - 26) SURYAWANSHI P G, BHANDARI V M, SOROKHAIBAM L G, et al. Solvent Degradation Studies Using Hydrodynamic Cavitation[J]. *Environmental Progress & Sustainable Energy*, 2018, 37(1): 295-304.
 - 27) Liao Z F, Tang C L. Theoretical Analysis and Experimental Study of the Pulsed Jet Device: Proceeding 4th US Water Jet Conference, Berkeley, 1984[C].